

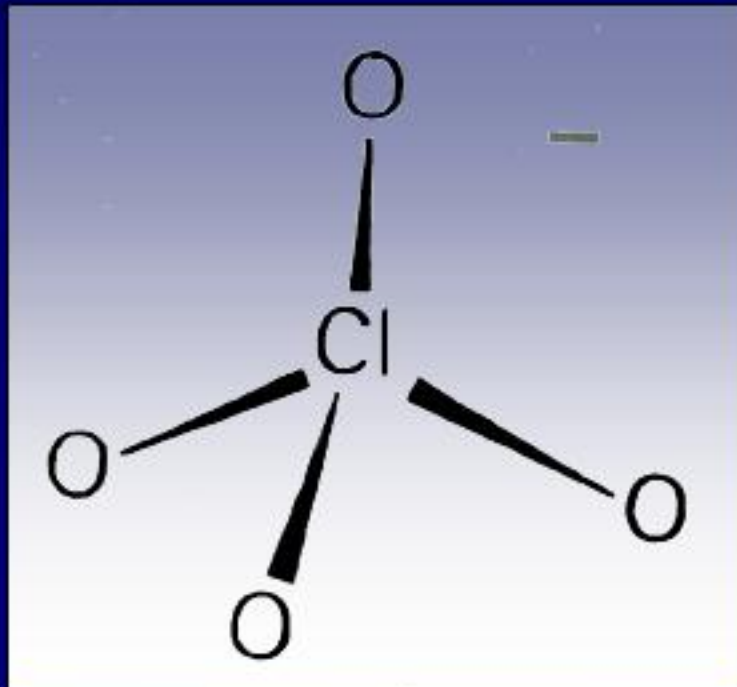
Perchlorate

An Integrated Approach to Reliable
Perchlorate Analyses

Agenda

- **Perchlorate Chemistry and Sources**
- **Current Methods for Testing**
- **Ion Chromatography**
- **LC/MS/MS**
- **An Integrated Approach**
- **A Success Story**
- **Summary and Conclusions**

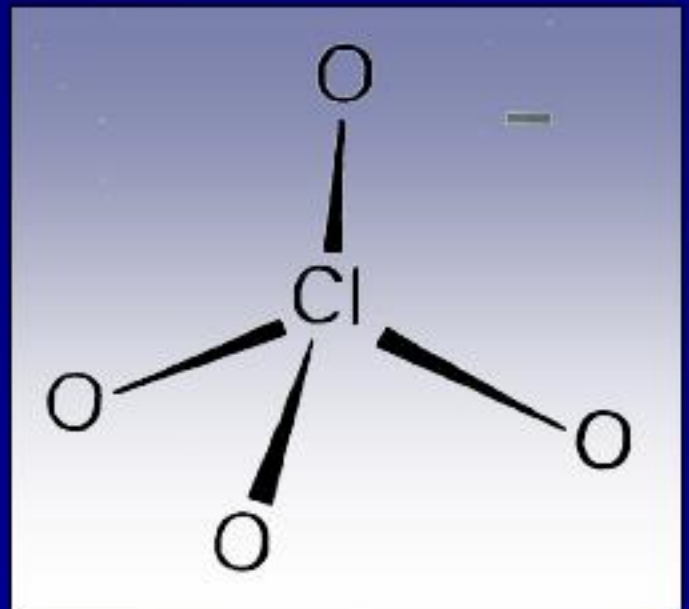
Perchlorate Ion



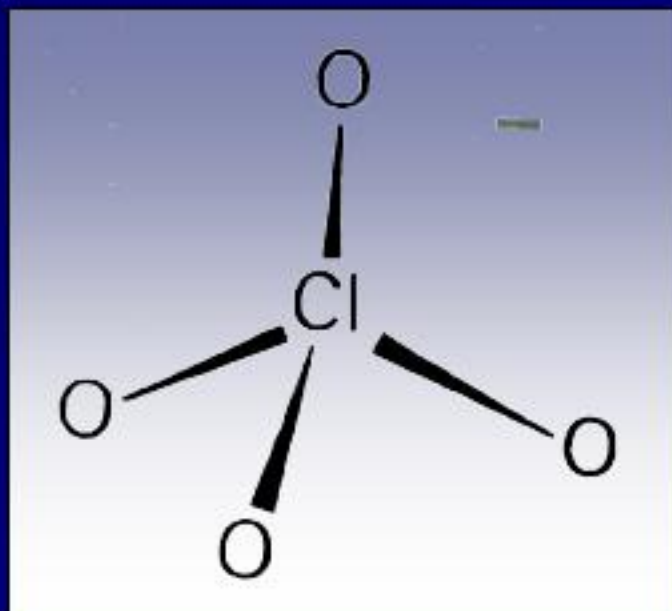
- Perchlorate is a compound containing one chlorine atom and four oxygen atoms.
- Perchlorate is an ion, meaning it carries an electrical charge, as indicated by the negative sign.

Natural Sources

- Chilean fertilizer deposits
- New Mexican potash
- Canadian potash
- Californian hanksite
- Bolivian playa crusts



Anthropogenic Sources



- Rocket fuel
- Fireworks
- High explosives
- Flares
- Herbicides
- Automobile airbags
- Tracer munitions
- Detergents ?

How is Perchlorate Measured?

- Other analytical methods that can detect perchlorate:
 - ◆ IC - 2nd column confirmation & pre-concentration
“Improved” Method 314.1
 - ◆ IC/MS - IC with mass spectrometer detection
Proposed Method 330.0
 - ◆ LC/MS - liquid chromatography with MS
detection
 - ◆ LC/MS/MS - Method 8321 Modified
Proposed Method 331.0

How is Perchlorate Measured?

- EPA Method 314.0
 - ◆ Ion chromatography (IC) with a conductivity detector
- Enhanced (Modified) 314.0
 - ◆ IC with a suppressed conductivity detection
- SW-846 Method 9058 (draft)
 - ◆ IC with a conductivity detector

Ion Chromatography

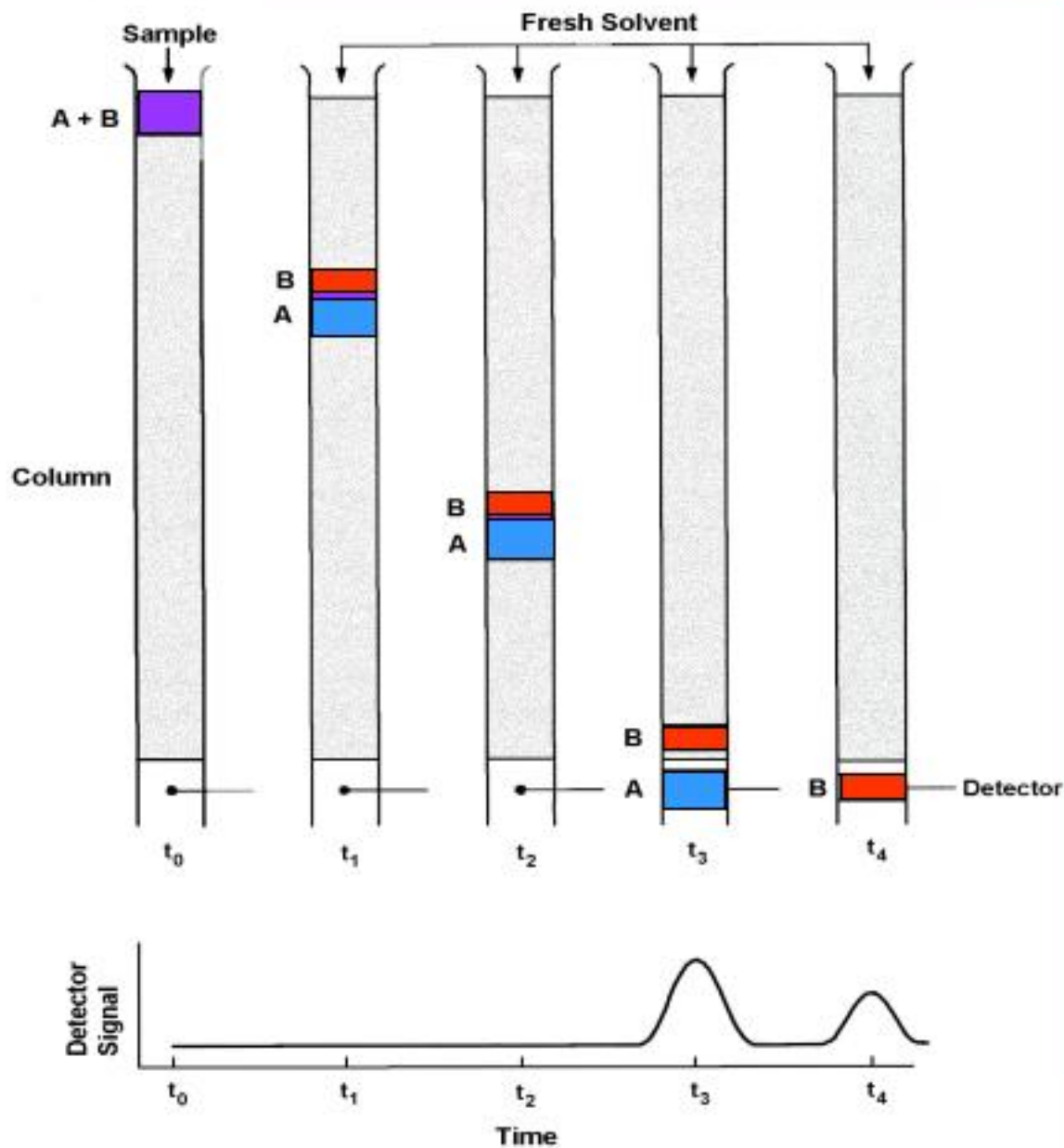
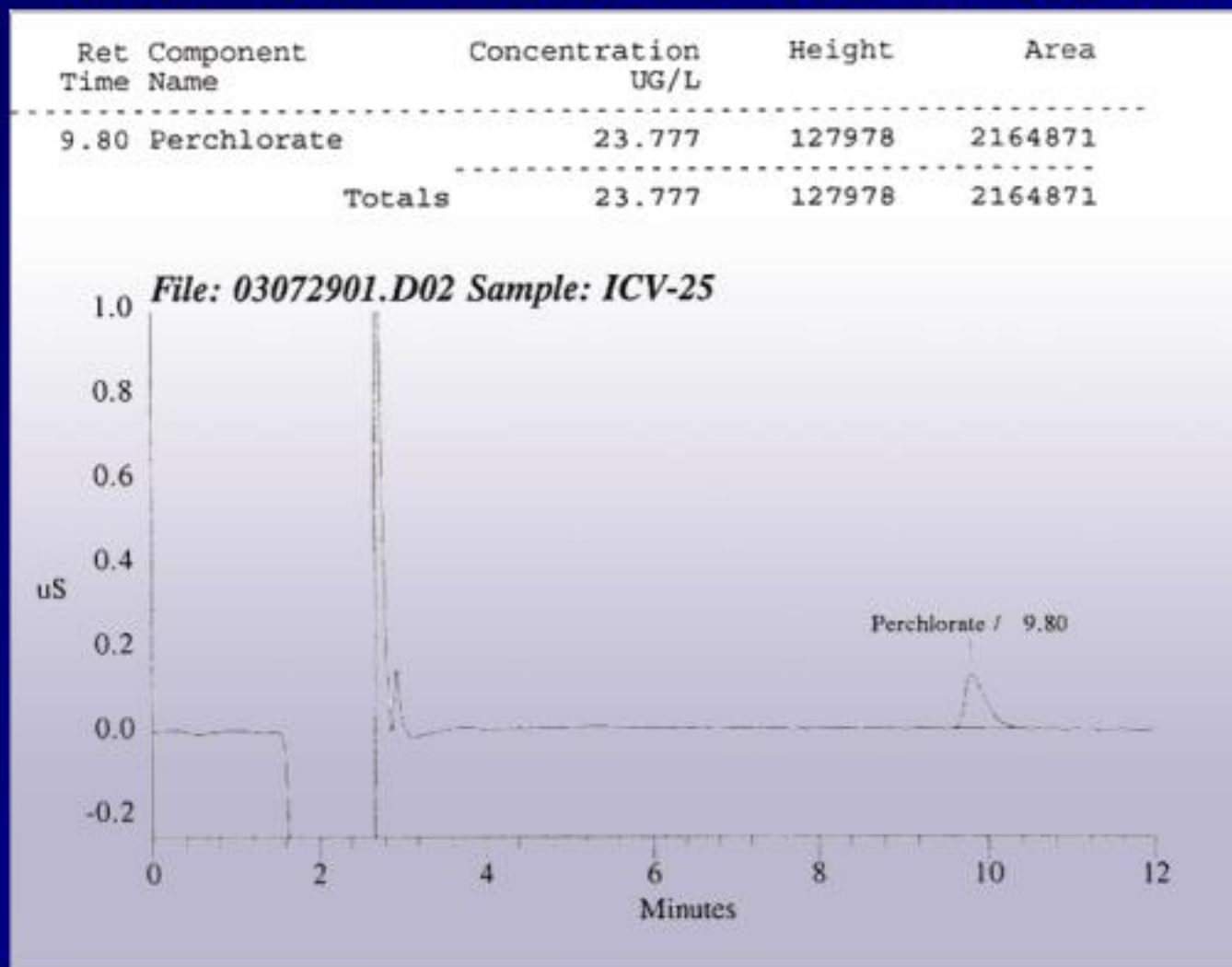


Diagram from *Fundamentals of Analytical Chemistry*, Skoog, 1988.

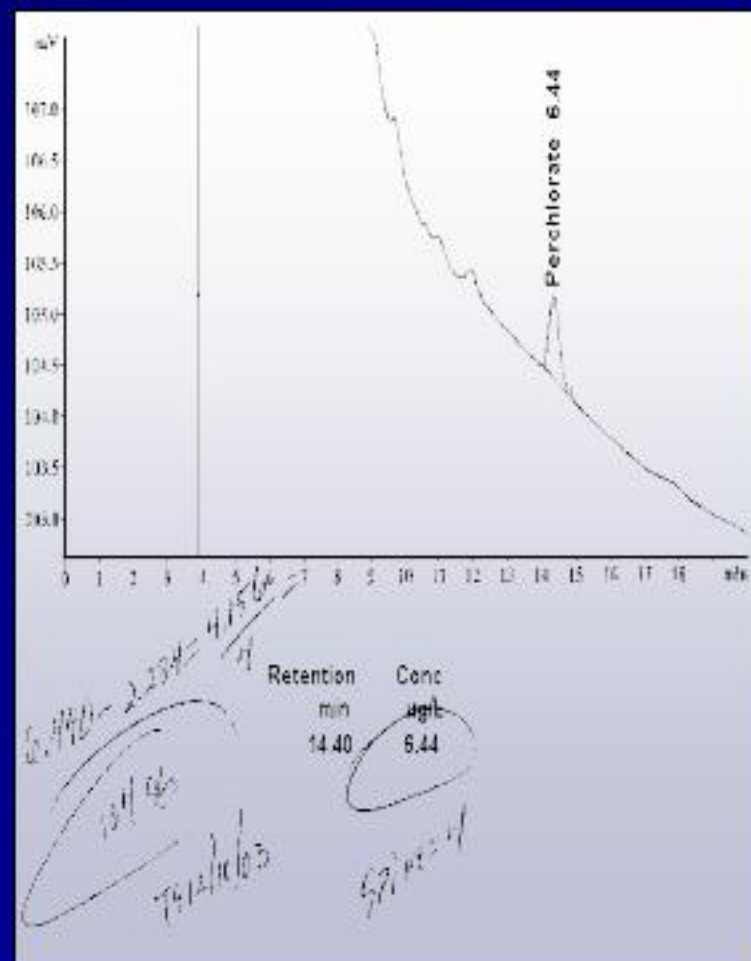
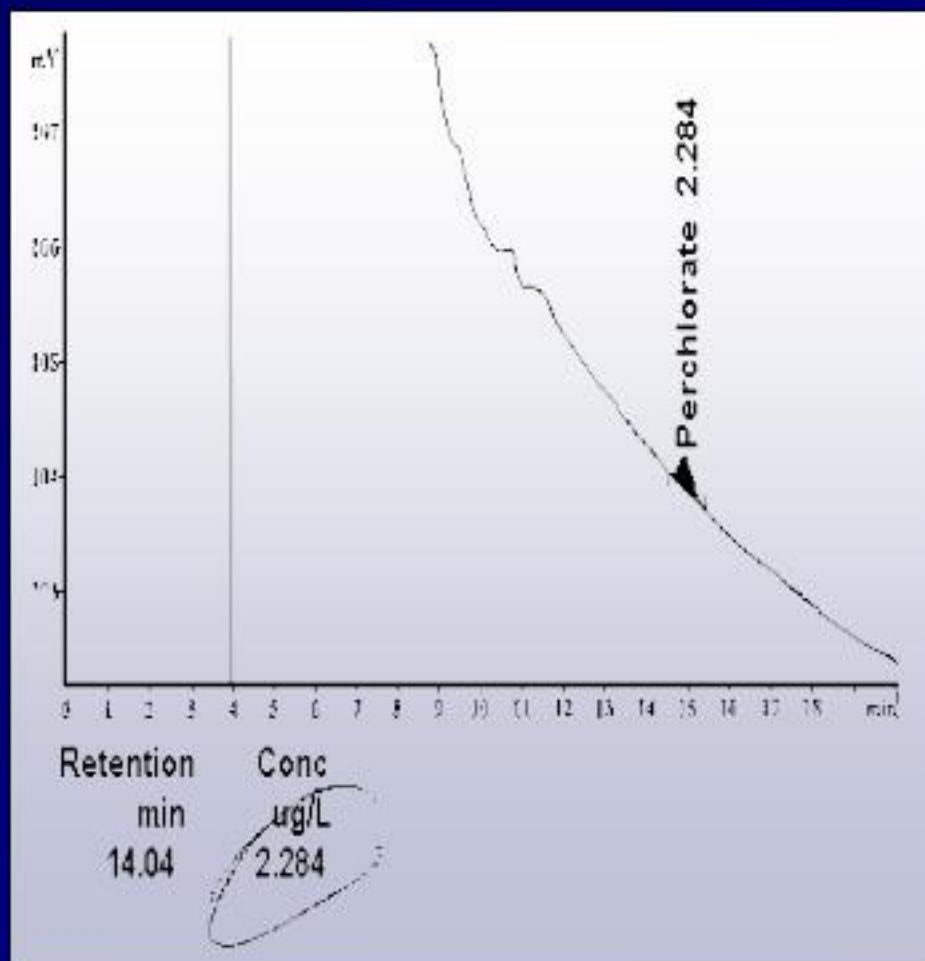
Identification of Perchlorate by IC

- 11.2.4 The width of the retention time window used to make identifications should be based upon measurements of actual retention time variations of standards measured over several days. Three times the standard deviation of retention time may be used as a suggested window size but the retention time window should not extend beyond $\pm 5\%$ of the retention time for perchlorate. The experience of the analyst should weigh heavily in the interpretation of these chromatograms.

IC Perchlorate Standard

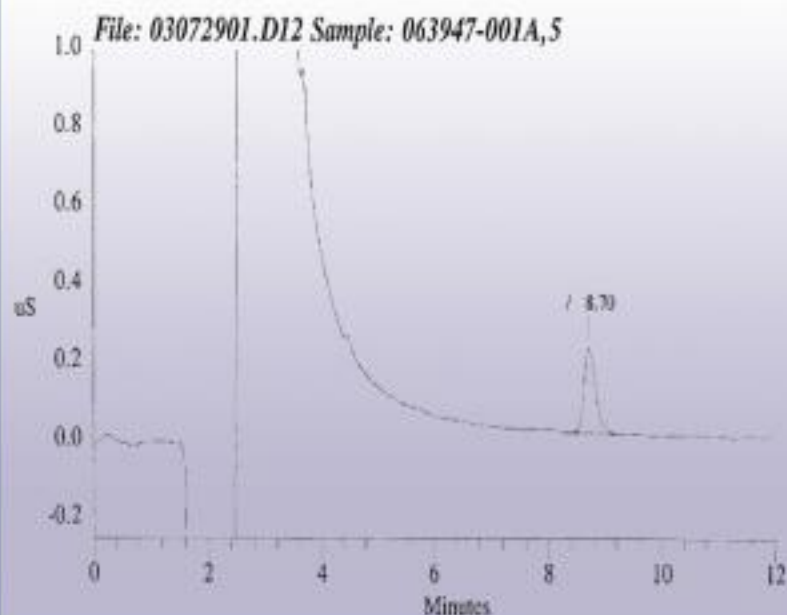


Matrix Spike Confirmation

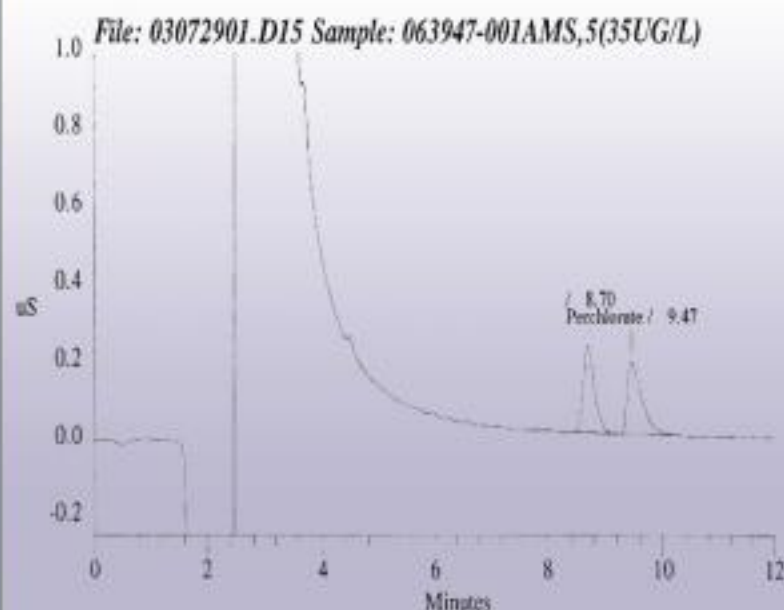


Matrix Spike Non-Confirmation

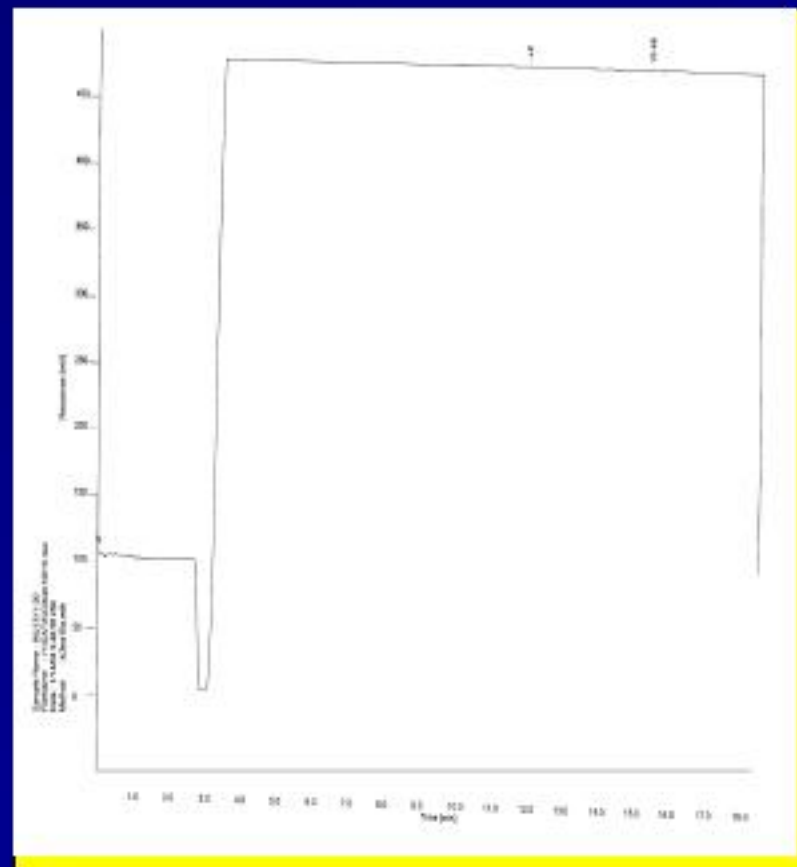
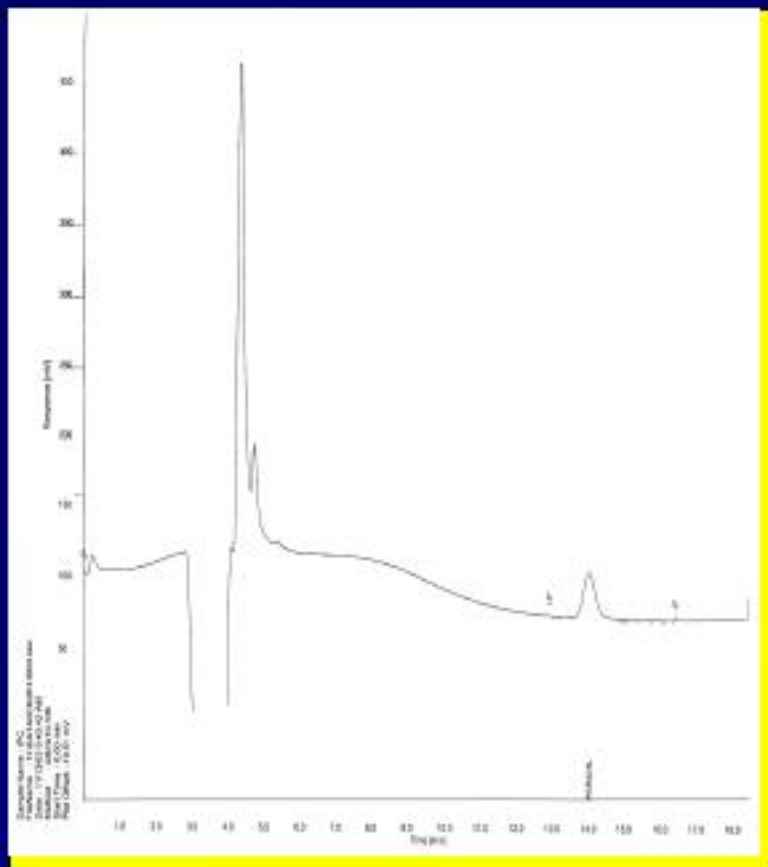
Ret Time	Component Name	Concentration UG/L	Height	Area
0.00	Perchlorate	0.000	0	0
Totals		0.000	0	0



Ret Time	Component Name	Concentration UG/L	Height	Area
9.47	Perchlorate	35.128	183878	3225568
Totals		35.128	183878	3225568



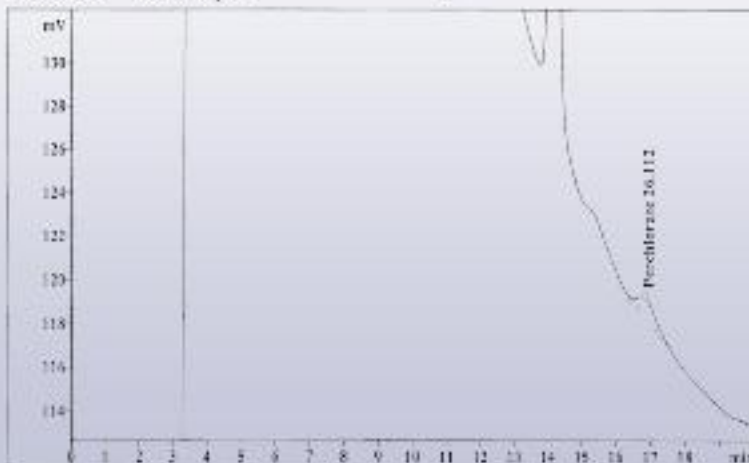
High Conductivity



Carry-Over

SAMPLE:
Vial number: 23
Volume: 1.0 µl
Dilution: 2.00
Amount: 1.0000
COLUMN: DX-AS16
Size: 4.0 x 250 mm
Number: 02362
Part size: 5.0 µm
ELUENT: 35 NaOH
Flow: 1.10 mL/min
Temperature: 35.0°C
Pressure: 2420.0 psi

*COND WAS 1450 µS/cm
NEEDED 2X DILUTION*

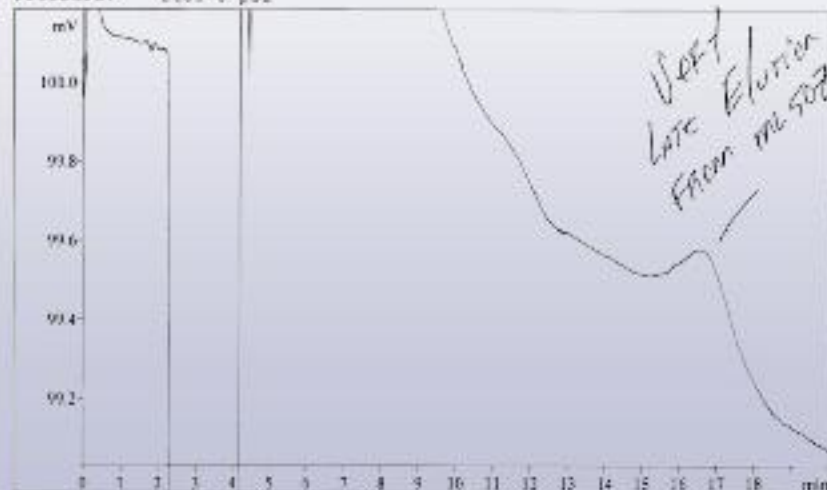


Quantitation method: Custom

No	Retention min	Height mV	Area µV*sec	Conc. µg/L	Name
1	16.112	0.90	31.655	15.112	Perchlorate

This report has been created by IC Net

SAMPLE:
Vial number: 26
Volume: 1.0 µl
Dilution: 1.00
Amount: 1.0000
COLUMN: DX-AS16
Size: 4.0 x 250 mm
Number: 02362
Part size: 5.0 µm
ELUENT: 35 NaOH
Flow: 1.10 mL/min
Temperature: 35.0°C
Pressure: 2400.0 psi



Quantitation method: Custom

No peaks

This report has been created by IC Net

Inconclusive Matrix Spike

SAMPLE:
Vial number: 41
Volume: 1.0 µl
Dilution: 1.00
Amount: 1.0000
COLUMN: DK-AS16
Size: 4.0 x 250 mm
Number: 02362
Part size: 5.0 µm
ELUENT: 35 NaOH
Flow: 1.10 mL/min
Temperature: 35.0°C
Pressure: 2400.0 psi

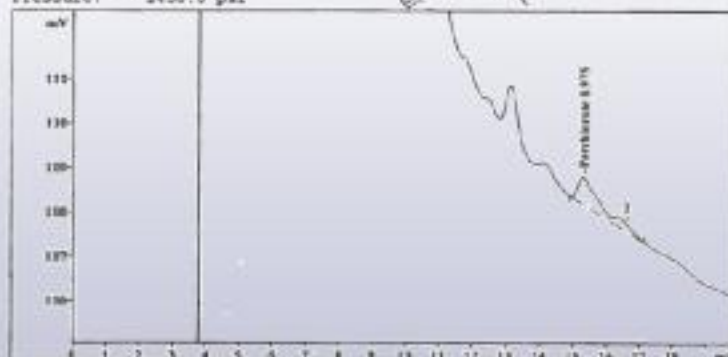


Quantitation method: Custom

No	Retention min	Height mV	Area mV*sec	Conc µg/l	Name
1	15.58	0.23	6.102	2.398	Perchlorate
2	16.83	0.16	4.884	0.000	
3	20.01	0.58	10.837	2.558	

*we have
the SPIKE*

SAMPLE:
Vial number: 43
Volume: 1.0 µl
Dilution: 1.00
Amount: 1.0000
COLUMN: DK-AS16
Size: 4.0 x 250 mm
Number: 02362
Part size: 5.0 µm
ELUENT: 35 NaOH
Flow: 1.10 mL/min
Temperature: 35.0°C
Pressure: 2400.0 psi



Quantitation method: Custom

No	Retention min	Height mV	Area mV*sec	Conc µg/l	Name
1	15.58	0.61	22.818	8.875	Perchlorate
2	16.83	0.16	4.724	0.000	
3	20.01	0.74	20.551	8.875	

*same as
but concentration is
increased by 4x (conclusion)*

Liquid Chromatography with Dual Mass Spectrometry

LC/MS/MS

- Leading edge technology
- DoD recommended
- Pending regulatory approval

Identification by LC/MS/MS

- LC separates perchlorate from other components in the same manner IC does
- Two mass spectrometers are used as detectors in place of the conductivity detector
- Unlike conductivity detectors, MS is a specific detector
 - ◆ Monitors molecular weights of parent (101/99 amu) and daughter ions (85/83 amu)

Results by LC/MS/MS

D3J160265-2 10X

F2N9F1AA

12:12:52

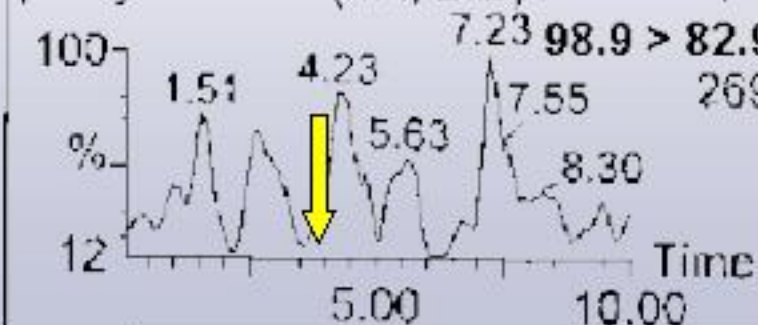
pe13j1716 Sm (Mn, 2x2) F1

100.9 > 84.9
266



pe13j1716 Sm (Mn, 2x2) F1

98.9 > 82.9
269



#	Name	RT	Area	Result ug(l or kg)
1	Perchlorate			

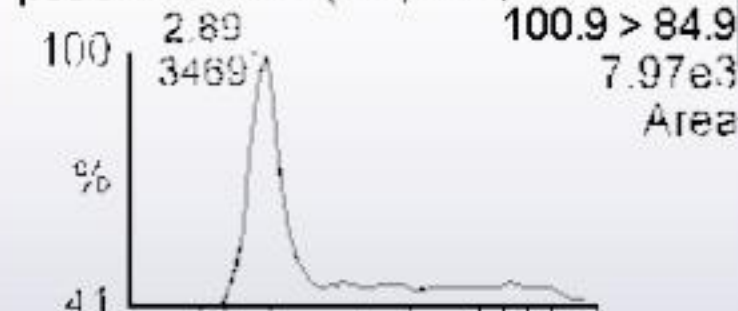
D3L300306-11 31-Dec-2003

F7GFP1AA

11:54:18

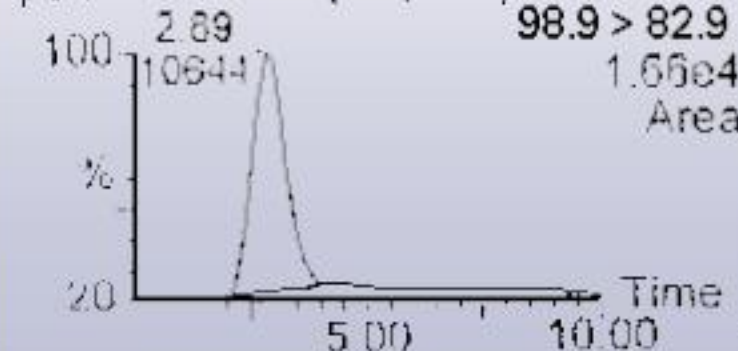
pe23j3125 Sm (Mn, 2x2) F1

100.9 > 84.9
7.97e3
Area



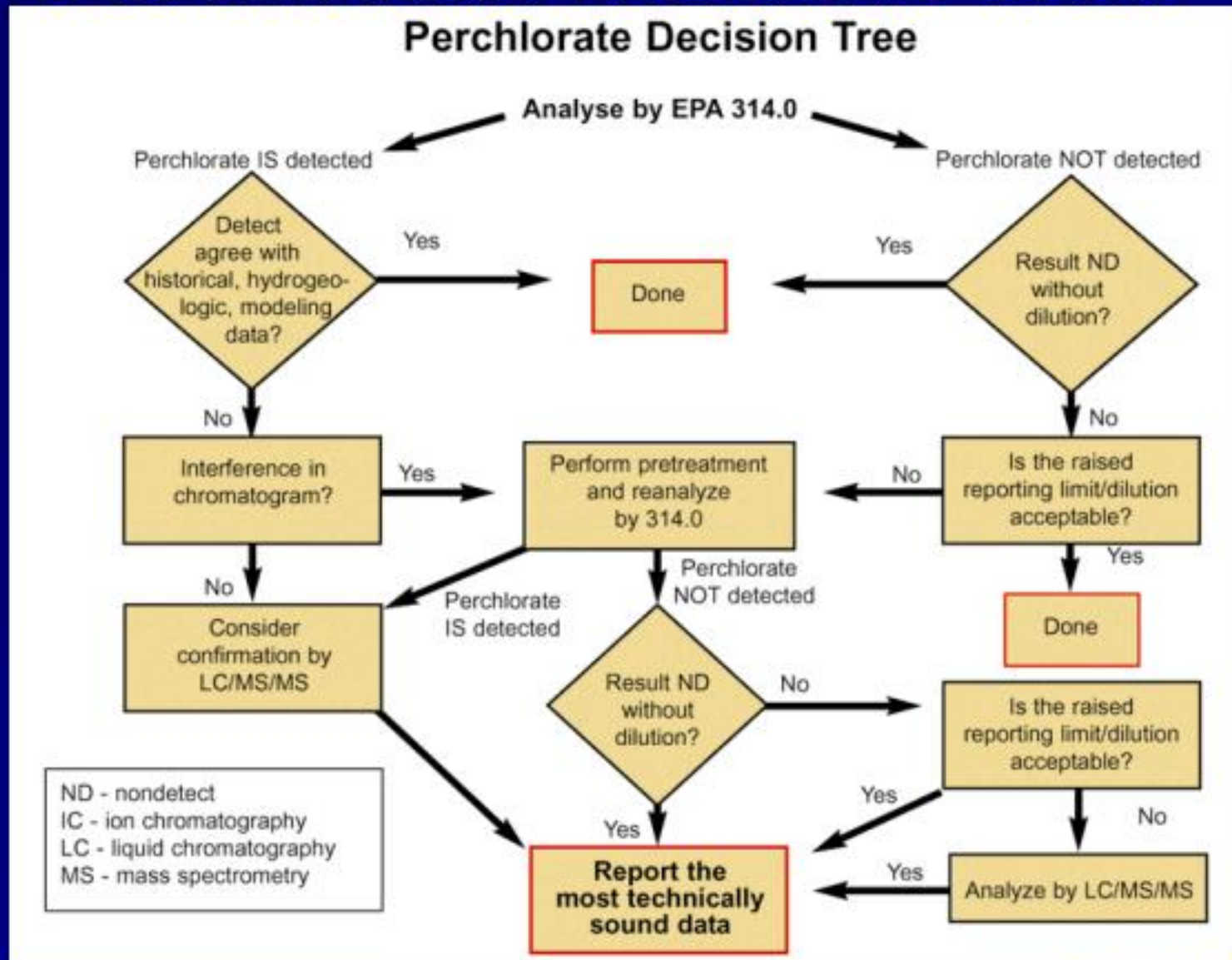
pe23j3125 Sm (Mn, 2x2) F1

98.9 > 82.9
1.66e4
Area



Name	RT	Result
Perchlorate	2.89	1.82

Perchlorate Decision Tree



Success Using the Perchlorate Decision Tree

- Approximately 1000 samples were collected and analyzed for perchlorate from August 2003 through January 2004
- 89 samples (about 9% of total) were pretreated and reanalyzed
- 39 samples (about 3% of total) were reanalyzed by LC/MS/MS - 24 detects and 15 non detects
- 10 perchlorate detects (about 0.4% of total) were confirmed by LC/MS/MS

Success Using the Perchlorate Decision Tree

■ Detects

- ◆ 16 showed definite signs of interference
- ◆ 2 had some interference
- ◆ 6 had little or no interference
- ◆ LC/MS/MS results equivalent to 314.0 for 9 confirmed results
- ◆ 314.0 had over-reported 1 confirmed result (46 vs. 15 ppb)

■ Non detects

- ◆ 6 reported at raised reporting limits
- ◆ 3 were highly suspicious
- ◆ 6 were chosen randomly

■ 58% false negative rate and 0% false positive rate

Success Using the Perchlorate Decision Tree

- Saved \$51,000 in analytical costs by not using LC/MS/MS exclusively
- Saved millions in remedial costs by not using 314.0 exclusively

Summary

- Why use 314.0?
 - ◆ Cost
 - ◆ Capacity
 - ◆ Availability
 - ◆ Promulgated Method
- When to use LC/MS/MS?
 - ◆ Definitive Data
 - ◆ Critical Samples
 - ◆ Small Projects

Thank You

Questions and Comments

Contact

Patti Meeks at 303.935.6505

patti.meeks@amec.com